

State Inventory Tool

User Guide for Estimating Carbon Dioxide, Nitrous Oxide, HFC, PFC, NF₃, and SF₆ Emissions from Industrial Processes

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ACKNOWLEDGEMENTS

The SIT was originally developed in conjunction with EPA’s state and local programs to automate the steps states would need to take in developing their own emission estimates in a manner that was consistent with prevailing national and state guidelines. The result was a user-friendly set of eleven modules that help users estimate greenhouse gas emissions and removals at the state level. CGS is supporting updates to the tool to facilitate continuity in state inventory development and improvement work at the state level.

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This section of the User Guide provides instruction on using the Industrial Processes (IP) module of the State Inventory Tool (SIT), and describes the methodology used for estimating carbon dioxide (CO₂), nitrous oxide (N₂O), HFC, PFC, NF₃, and SF₆ emissions from industrial processes at the state level.

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1.1 MODULE OVERVIEW

This User Guide accompanies and explains the IP module of the SIT.

Because most state inventories developed today rely heavily on the SIT, User Guides have been developed for each of the SIT modules. These modules contain largely updated methodologies that are, for the most part, consistent with the Inventory of U.S. Greenhouse Gas Emissions and Sinks (EPA 2025) and related disaggregated data by state. Users can refer to the chapters and annexes of the U.S. Inventory to obtain additional information not found in the SIT or in the companion User Guide. Future updates will be consistent with the Center for Global Sustainability at University of Maryland's Greenhouse Gas Inventory and Analysis for the United States (GHGIA) (Desai et al. 2026).

In recent years, EPA also published an annual Inventory of U.S. Greenhouse Gas Emissions and Sinks by State (EPA 2024a) to make available consistent state-level GHG data for all states for use by states, researchers, and the general public. There will be differences between the latest EPA developed state-level estimates and inventory estimates developed by states using the SIT or other tools. Inventories compiled by states may differ for several reasons, and differences do not necessarily mean that one set of estimates is more accurate, or "correct". The CGS plans to continue EPA's initiative to publish an updated annual crosswalk guide to facilitate understanding methodological differences between methods in the current GHGIA, its disaggregation by state and SIT source categories, methods, and data.

The IP module calculates carbon dioxide (CO₂), nitrous oxide (N₂O), hydrofluorocarbon (HFC), perfluorocarbon (PFC), nitrogen trifluoride (NF₃), and sulfur hexafluoride (SF₆) emissions from the IP sectors shown in Table 1. While the module provides default data for each sector (depending on availability), if you have access to a more comprehensive data source, it should be used in place of the default data. If using outside data sources, or for a more thorough understanding of the tool, please refer to the following discussion for data requirements and methodology.

1.1.1 Data Requirements

To calculate CO₂, N₂O, HFC, PFC, NF₃, and SF₆ emissions from IP, general activity data on various IP sectors are required. A complete list of the activity data and emission factors necessary to run the IP module is provided in Table 1.

Table 1. IP Sectors, Data Requirements, and Gases Emitted

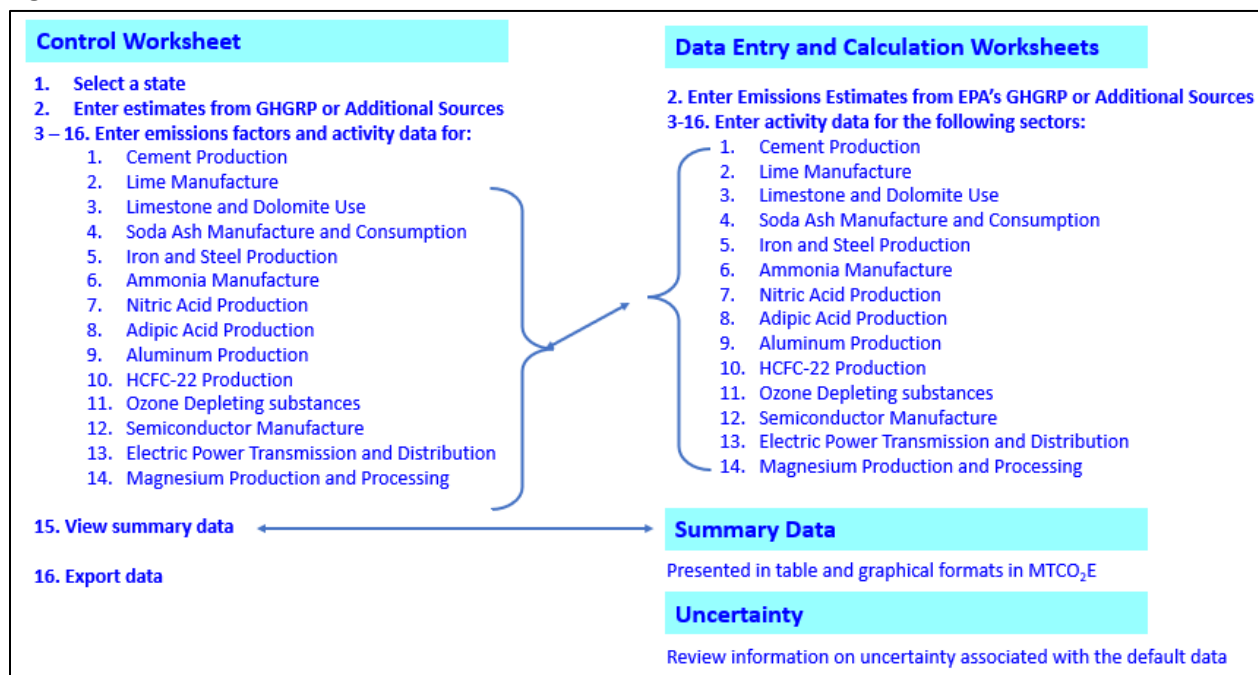
Module Worksheet	Data Required	Gas(es)
Cement Production	Emission factors and production data for clinker and cement kiln dust (CKD)	CO ₂
Lime Manufacture	Emission factors and production data for high-calcium lime, and dolomitic lime	
Limestone and Dolomite Use	Emission factors and consumption data for limestone, dolomite, and magnesium produced from dolomite	
Soda Ash Manufacture and Consumption	Emission factors and consumption data for manufacture and consumption of soda ash	
Iron and Steel Production	Emission factors and production data for Basic Oxygen Furnace (BOF) at Integrated Mill with Coke Ovens, Basic Oxygen Furnace (BOF) at Integrated Mill without Coke Ovens, Electric Arc Furnace (EAF), and Open Hearth Furnace (OHF)	
Ammonia Manufacture	Emission factors and production and consumption data for ammonia production, and urea consumption	
Nitric Acid Production	Emission factor, production data, and Percent N ₂ O Released after Pollution Control for nitric acid production	N ₂ O
Adipic Acid Production	Emission factor, production data, and Percent N ₂ O Released after Pollution Control for adipic acid production	
Aluminum Production	Emission factors for Prebake and Söderberg technologies and aluminum production data by technology	CO ₂ and PFC
HCFC-22 Production	Emission factor and production data for HCFC-22 production	HFC, PFC, NF ₃ , and SF ₆
Consumption of Substitutes for Ozone-Depleting Substances (ODS)	No input data required*	
Semiconductor Manufacture	No input data required*	
Electric Power Transmission and Distribution	Emission factor and SF ₆ consumption data for electric power transmission and distribution	
Magnesium Production and Processing	Emission factor and consumption data for primary production, secondary production, and casting	

*According to inventory guidance, emissions of HFCs, PFCs, NF₃, and SF₆ from ODS substitutes and semiconductor manufacture can be estimated by apportioning national or regional emissions to each state based on population. Because this tool pulls in national emissions apportioned based on state population, the emission factors and activity data for these sources are not required.

1.1.2 Tool Layout

Because there are multiple steps to complete within the IP module, it is important to understand the module's overall design. The layout of the IP module is presented in Figure 1.

Figure 1. Flow of Information in the IP Module*



* According to inventory guidance, emissions of HFCs, PFCs, NF₃, and SF₆ from ODS substitutes and semiconductor production can be estimated by apportioning national emissions to each state based on population. Because this tool pulls in national emissions apportioned based on state population, no emission factors need to be entered for these sources.

1.2 METHODOLOGY

This section provides a guide to using the IP module of the SIT to estimate CO₂, N₂O, NF₃, HFC, PFC, and SF₆ emissions from IP. The sectors included in the IP module are cement production, lime manufacture, limestone and dolomite use, soda ash manufacture and consumption, iron and steel production, ammonia manufacture, nitric and adipic acid production, aluminum production, HCFC-22 production, consumption of substitutes for ozone depleting substances, semiconductor manufacture, electric power transmission and distribution, and magnesium production and processing. Because the methodology varies by sector, they are discussed separately and specific examples for each sector are provided.

The IP module follows the general methodology from the National Inventory Report (NIR) GHG Inventory by U.S. State. However, SIT does not include all source categories (e.g., lead production, carbide production and consumption) or have default data available for all source categories or years (e.g., adipic acid, lime production, nitric acid production).

This User Guide provides an overview of the estimation methodology used in the IP module by walking through the following steps: (1) select a state; (2) enter available data aggregated for EPA's Greenhouse Gas Reporting Program and additional emission sources; (3) enter emission factors and activity data for cement production; (4) enter emission factors and activity data for lime manufacture; (5) enter emission factors and activity data for limestone and dolomite use; (6) enter emission factors and activity data for soda ash manufacture and consumption; (7) enter emission factors and activity data for iron and steel production; (8) enter emission factors and activity data for ammonia manufacture; (9) enter emission factors and activity data for nitric acid production; (10) enter emission factors and activity data for adipic acid production; (11) enter emission factors and activity data for aluminum production; (12) enter emission factors and activity data for HCFC-22 production; (13) review sector worksheet for consumption of substitutes for ozone depleting substances; (14) review sector worksheet for semiconductor manufacture; (15) enter emission factors and activity data for electric power transmission and distribution; (16) complete control and sector worksheets for magnesium production and processing; and (17) review summary information. The general equations used to calculate CO₂, N₂O, and HFC, PFC, NF₃, and SF₆ emissions from IP are shown in the discussion of each specific sector.

Step (1) Select a State

To begin, select the state you are interested in evaluating. By selecting a state, the rest of the tool will automatically reset to reflect the appropriate state default data and assumptions for use in subsequent steps of the tool.

Step (2) Enter Emission Estimates for Facilities Reporting to EPA's Greenhouse Gas Reporting Program or Additional Emission Sources (OPTIONAL)

Additional data, such as emissions and activity data, have historically been available through EPA's Greenhouse Gas Reporting Program (GHGRP) starting in 2010. You have the option to rely on this data to estimate emissions for select source categories. Future data (i.e., 2025 onward) will need to be sourced from the facilities themselves or a different data source.

The GHGRP historically required reporting of greenhouse gas (GHG) data and other relevant information from large sources and suppliers in the United States. EPA's reporting threshold for the GHGRP was for facilities that emit a total of 25,000 MTCO₂E per year. However, in some instances, source categories were "all in" or do not have a reporting threshold have been identified (i.e., all facilities in operation for that source category need to report GHG emissions to EPA), which simplifies the determination process for GHGRP reporting applicability.

Cement production, lime manufacturing, soda ash manufacturing, ammonia manufacturing, nitric acid production, adipic acid production, aluminum production, and HCFC-22 production/HFC-23 destruction were the industrial process source categories that are "all in". Information about the GHGRP can be accessed at: <https://www.epa.gov/ghgreporting>.

If you would like to incorporate GHGRP data into your state’s GHG inventory for the source categories identified above, follow the steps outlined below.

1. Use EPA’s Facility-Level Information on GreenHouse Gases Tool (FLIGHT) to determine if your state has applicable emissions from the GHGRP to incorporate in your GHG inventory.
 - a. Go to <https://ghgdata.epa.gov/ghgp/main.do>.
 - b. Select your state on the initial screen pop up or on the top left-hand portion of the webpage.
 - c. At the bottom of the webpage (under the map), remove checks from all sectors except “Chemicals”, “Minerals”, or “Metals”.¹
 - d. Hover over the gear icon to the top right of the selected sector, and only check the source category of interest to determine if applicable emissions are available. (Note: Only select one source category at a time to view emissions from that source.)
 - e. Select the button on the left side of the webpage labeled "APPLY SEARCH".
 - f. View if you have emissions for the selected source category in your state.
 - g. Repeat steps 1a-1f for each source category listed above to determine if your state has applicable emissions from the GHGRP to incorporate in your GHG inventory.

Note: Do not pull emissions directly from FLIGHT to incorporate into the IP Module, as they include both process and stationary combustion emissions, and this will result in double counting in your GHG inventory between the IP and Stationary Combustion Modules.

2. Once applicability is determined, use EPA’s Envirofacts Customized Search to access and view process emissions for most source categories. (For adipic acid, HCFC-22 production/HFC-23 destruction, lime manufacturing [CEMS reporters only], and soda ash manufacturing see Step 3.)
 - a. Go to <https://www.epa.gov/enviro/greenhouse-gas-customized-search>.
 - b. Scroll down and select the relevant source category.
 - c. At the bottom of the next page, select “Step 2: Retrieve Tables for Selected Subjects”.
 - d. On the next page, select the table you would like to view by clicking the radio button to the left of the table title, and then select “Step 3: Select Columns”. (Key words to look for when selecting the table to view include “subpart-level”, and/or “emissions”. A common table name you may select is “Subpart_Level_Information”.)
 - e. On the next page, select the Columns you would like to view by checking the box to the left, and then select “Step 4: Enter Search Criteria”. (Key words to look for when selecting columns include “emissions”, “specified GHG”, or “facility-wide”. A common column name you may select is “Greenhouse Gas Quantity”.)
 - f. On the next page, enter your state’s abbreviation in the State Abbreviation form, and the relevant reporting year, then select “Search Database” or “Output to CSV File” at the

¹ “Chemicals” include ammonia manufacturing, nitric acid production, adipic acid production, and HCFC-22 production/HFC-23 destruction. “Minerals” include cement production, lime manufacturing, and soda ash manufacturing. “Metals” include aluminum production.

g. Sum the relevant emissions data in the CSV file and then transfer the CSV results to the “GHGRP Data Input” tab in the relevant worksheet cells.

3. For select source categories (adipic acid, HCFC-22 production/HFC-23 destruction, lime manufacturing [CEMS reporters only], and soda ash manufacturing), detailed process emissions data must be accessed through an EPA-published spreadsheet available here: <https://www.epa.gov/ghgreporting/ghg-reporting-program-data-sets>, in the file titled “Subpart E, O, S-CEMS, BB, CC, LL, RR Data Set”. Transfer the relevant emissions data to the “GHGRP Data Input” tab in the relevant worksheet cells.

Users may also use this worksheet to input data about additional emissions sources that are not included in this module or any other module. Additional emissions sources should be reported in MTCO₂e. Figure 2 shows two options for entering data into this worksheet (please select only one):

Option A: Add up to ten additional emission sources in column B, lines 30-39. Enter the emissions for each gas in MMTCO₂e for each year 1990-2025 in the yellow cells.

Option B: Enter the total emissions for CO₂, N₂O, HFCs, PFCs, SF₆, and NF₃ (lines 53-58) for all other emissions sources for each year 1990-2025 in the yellow cells in MMTCO₂e.

[illegible]

There are several categories included in the U.S. Inventory that are not available or only partially available in SIT, including production of: lime, glass, nitric acid, adipic acid, carbide, titanium dioxide, petrochemicals, phosphoric acid, ferroalloys, iron and steel and metallurgical coke, aluminum,

magnesium, lead, and zinc. The U.S. Inventory also includes carbide consumption, carbon dioxide consumption, and N₂O from product uses. For emission sources that do not have any default data or are not accounted for within the State Inventory Tool, state-level data are available here:

<https://www.epa.gov/ghgemissions/state-ghg-emissions-and-removals>

Step (3) Enter Emission Factors and Activity Data for Cement Production

Control Worksheet

The second step for the control worksheet is to either select the default data provided or to enter user-specified data that will be used throughout the tool. To proceed with the default data, select the “Clear/Select All” button for each sector on the control worksheet or check the individual default box directly to the right of specific yellow input cells. See Figure 3 for locations of the “Clear/Select All” buttons, individual default check boxes, and yellow input cells. Note that this number can be overwritten if you later discover that the data for your state differ from the default data provided by the tool. To enter user-specified inputs, enter data directly into the yellow input cells. If the user-specific inputs do not match the default data in the control worksheet (i.e., the default value is overwritten), the text will appear red. Information requirements on the control worksheet for each sector are discussed separately below.

Figure 3. Control Worksheet for the IP Module

State Inventory Tool - Industrial Processes Module

1. Choose a State: Colorado
This is very important – it selects the correct default variables for your state.

2. Enter emission estimates for facilities reporting to EPA's Greenhouse Gas Reporting Program

3.- 17. Enter emission factors and proceed to the sector worksheet to complete activity data for the following industrial processes:
Either type in the value/percentage or click the default box
CO₂ Emission Factors in Units of CO₂ Emitted per Unit of Production or Consumption

Process	Default Factor	Input Cell	Use the Default? (Check for Yes)	Action
3. Cement Production				
Clinker	0.507	Metric Tons CO ₂ Emitted / Metric Ton of Clinker Produced	☐	Go to the Cement Sheet
Cement Kiln Dust (CKD)	0.020	Metric Tons CKD CO ₂ Emitted / Metric Ton of Clinker CO ₂ Emitted	☐	Go to the Cement Sheet
4. Lime Manufacture				
High-Calcium Lime	0.75	Metric Tons CO ₂ Emitted / Metric Ton High-Calcium Lime Produced	☐	Go to the Lime Sheet
Dolomitic Lime	0.87	Metric Tons CO ₂ Emitted / Metric Ton Dolomitic Lime Produced	☐	Go to the Lime Sheet
5. Limestone and Dolomite Use				
Limestone	0.440	Metric Tons CO ₂ Emitted / Metric Ton Limestone (Calcite)	☐	Go to the Limestone Sheet
Dolomite	0.494	Metric Tons CO ₂ Emitted / Metric Ton Dolomite	☐	Go to the Limestone Sheet
Magnesium Produced from Dolomite	1.797	Metric Tons CO ₂ Emitted / Metric Ton Magnesium Produced	☐	Go to the Soda Ash Sheet
6. Soda Ash Manufacture and Consumption				
Manufacture	0.097	Metric Tons CO ₂ Emitted / Metric Ton Soda Ash Produced	☐	Go to the Soda Ash Sheet
Consumption	0.415	Metric Tons CO ₂ Emitted / Metric Ton Soda Ash	☐	Go to the Soda Ash Sheet
7. Iron and Steel Production				
Basic Oxygen Furnace (BOF) at Integrated Mill with Coke Ovens	1.720	Metric Tons CO ₂ Emitted / Metric Ton Crude Steel Produced	☐	Go to the Iron & Steel Sheet
Basic Oxygen Furnace (BOF) at Integrated Mill without Coke Ovens	1.460	Metric Tons CO ₂ Emitted / Metric Ton Crude Steel Produced	☐	Go to the Iron & Steel Sheet
Electric Arc Furnace (EAF)	0.080	Metric Tons CO ₂ Emitted / Metric Ton Crude Steel Produced	☐	Go to the Iron & Steel Sheet
QHF	1.720	Metric Tons CO ₂ Emitted / Metric Ton Crude Steel Produced	☐	Go to the Iron & Steel Sheet
8. Ammonia Manufacture				
Ammonia Production	1.2	Metric Tons CO ₂ Emitted / Metric Ton NH ₃ Produced	☐	Go to the Ammonia & Urea Sheet
Urea Production	0.73	Metric Tons CO ₂ Emitted / Metric Ton NH ₃ Produced	☐	Go to the Ammonia & Urea Sheet
9. Aluminum Production				
Aluminum Production - Prebake	0.436	Metric Tons CO ₂ Emitted / Metric Ton Aluminum Produced	☐	Go to the Aluminum Sheet
Aluminum Production - Soderberg	0.464	Metric Tons CO ₂ Emitted / Metric Ton Aluminum Produced	☐	Go to the Aluminum Sheet

The first type of required data in the control worksheet is emission factors for clinker, and cement kiln dust used in cement production. CO₂ emissions from cement production consist of emissions produced during the cement clinker production processes and are in units of metric tons of CO₂ released per metric ton of clinker or cement kiln dust produced. Emissions from the production of masonry cement are accounted for in Lime emissions estimates.

Cement Production Sector Worksheet

The activity data required to populate the blue cells in the cement production worksheet are metric tons of clinker produced annually, as shown in Figure 4. Select “Check All Boxes” if you would like to use default data provided in the IP module. Activity data for cement production by state is available from USGS (2026a). CO₂ is created when calcium carbonate (CaCO₃) is heated in a cement kiln to form lime (calcium oxide or CaO) and CO₂. This process is known as calcination or calcining.

Cement clinker emissions are calculated by multiplying the clinker production quantity by the emission factor entered on the control worksheet and adding the product to the emissions from cement kiln dust (a by-product of cement clinker production). The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E). Equation 1 shows this calculation for CO₂ emissions from cement production.

Equation 1. Emission Equation for Cement Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Production (metric tons)} \times \text{Emission Factor (t CO}_2\text{/t production)} + \text{Emissions from Cement Kiln Dust (Metric tons CO}_2\text{)}$$

Figure 4. Example of Activity Data Applied in the Cement Production Worksheet

State Inventory Tool - Industrial Processes Module

2. Cement Production in Colorado

Click here to find where these data are available.

Select All Defaults

Return to Control Sheet

Check All Boxes

Clear All Data

	Production (Metric Tons)	Emission Factor (t CO ₂ /t production)	Emissions (Metric Tons CO ₂)	Emissions from Cement Kiln Dust (Metric Tons CO ₂)	Emissions (MTCE)	Emissions (MTCO ₂ E)	
1990 Clinker	815,888	0.0000	-	-	-	-	Default Production Data?
1991 Clinker	847,510	0.0000	-	-	-	-	Default Production Data?
1992 Clinker	894,457	0.0000	-	-	-	-	Default Production Data?
1993 Clinker	948,500	0.0000	-	-	-	-	Default Production Data?
1994 Clinker	949,000	0.0000	-	-	-	-	Default Production Data?
1995 Clinker	920,000	0.0000	-	-	-	-	Default Production Data?
1996 Clinker	914,000	0.0000	-	-	-	-	Default Production Data?
1997 Clinker	982,000	0.0000	-	-	-	-	Default Production Data?
1998 Clinker	979,000	0.0000	-	-	-	-	Default Production Data?
1999 Clinker	998,000	0.0000	-	-	-	-	Default Production Data?
2000 Clinker	1,070,888	0.0000	-	-	-	-	Default Production Data?
2001 Clinker	996,500	0.0000	-	-	-	-	Default Production Data?
2002 Clinker	958,000	0.0000	-	-	-	-	Default Production Data?
2003 Clinker	1,175,000	0.0000	-	-	-	-	Default Production Data?
2004 Clinker	1,242,897	0.0000	-	-	-	-	Default Production Data?
2005 Clinker	1,204,500	0.0000	-	-	-	-	Default Production Data?

Required Data Input Cells

Emission Factors

Control Cement Lime Limestone Soda Ash Iron & Steel Ammonia & Urea Nitric Acid Aluminum HCFC-22 ODS Semiconductor Electric Power Ma

Step (4) Enter Emission Factors and Activity Data for Lime Manufacture

Control Worksheet

The emission factors for high-calcium lime and dolomitic lime manufacture are the next required inputs on the control worksheet. Lime is manufactured by heating limestone (mostly CaCO_3) in a kiln, creating CaO and CO_2 . The IP module estimates these CO_2 emissions from two types of lime: high-calcium lime and dolomitic lime production.

Lime Manufacture Sector Worksheet

Production data for high-calcium dolomite and dolomitic lime, and the amount of these used in sugar refining and precipitated calcium carbonate are required inputs in the blue cells of the lime manufacture worksheet as shown in Figure 5. Activity data for lime manufacture by state is available from USGS (2026b).

Before entering the production of high-calcium and dolomitic lime, you must correct for the water content of hydrated lime. The water content can be assumed to be 24.3 percent for high-calcium hydrated lime and 27.3 percent for dolomitic lime. To correct for the water content of hydrated lime, multiply the production data for high-calcium hydrated lime and dolomitic hydrated lime by their respective percentages of *dry* lime to find the corrected production numbers for both varieties of hydrated lime. An example of this correction for high-calcium lime is shown in Equation 2.

Equation 2. Example Calculation for Hydrated Lime Correction

$$\text{Corrected Lime Content of High-Calcium Hydrated Lime (metric tons)} = \text{High-Calcium Hydrated Lime Production (metric tons)} \times (1 - 0.24 \text{ metric tons water/metric ton high-calcium hydrated lime})$$

To calculate emissions from this source, the production quantity of each lime type is multiplied by its respective emission factor from the control worksheet. Because lime used in sugar refining and precipitated calcium carbonate production results in the reabsorption of atmospheric CO_2 , carbon absorbed from these uses is subtracted from gross emissions. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO_2E). Equation 3 shows this calculation for CO_2 emissions from cement production.

Equation 3. Emission Equation for Lime Manufacture

$$\text{Emissions (MTCO}_2\text{E)} = [\text{Production (metric tons)} - \text{Sugar Refining and Precipitated Calcium Carbonate Production (metric tons)} \times \text{CO}_2 \text{ Reabsorption Factor (80\%)}] \times \text{Emission Factor (MT CO}_2\text{/MT production)}$$

Figure 5. Example of Activity Data Applied in the Lime Manufacture Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options

Type a question for help

3. Lime Manufacture in Colorado

Click here to find where these data are available.

Emissions from lime manufacture consist of emissions from high-calcium and dolomitic lime production. The production quantity of each lime type is multiplied by its respective emission factor. Because lime used in sugar refining and precipitated calcium carbonate production results in the reabsorption of atmospheric CO₂, carbon absorbed from these uses is subtracted from gross emissions. The emissions are then converted to metric tons of carbon equivalents (MTCE) and from metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Check All Boxes

Clear All Data

		Production (Metric Tons)	Use in Sugar Refining and Precipitated Calcium Carbonate Production (Metric Tons)	CO ₂ Reabsorption Factor	Emission Factor (t CO ₂ /t production)	Emissions (MTCE)	Emissions (MTCO ₂ E)	
1991	High-Calcium Lime	69,299		80%	0.7500	14,175	51,974	✓ Default Production Data
	Dolomitic Lime	15,372		80%	0.8700	3,647	13,374	
1992	High-Calcium Lime	99,106		80%	0.7500	20,272	74,329	✓ Default Production Data
	Dolomitic Lime	21,853		80%	0.8700	5,185	19,012	
1993	High-Calcium Lime	280,964		80%	0.7500	57,470	210,723	✓ Default Production Data
	Dolomitic Lime	62,012		80%	0.8700	14,720	53,972	
1994	High-Calcium Lime	264,392		80%	0.7500	54,204	199,749	✓ Default Production Data
	Dolomitic Lime	58,001		80%	0.8700	13,762	50,461	
1995	High-Calcium Lime	106,719		80%	0.7500	21,829	80,039	✓ Default Production Data
	Dolomitic Lime	23,281		80%	0.8700	5,524	20,254	
1996	High-Calcium Lime	95,313		80%	0.7500	19,496	71,485	✓ Default Production Data
	Dolomitic Lime	20,697		80%	0.8700	4,908	17,997	
1997	High-Calcium Lime	92,588		80%	0.7500	18,939	69,441	✓ Default Production Data
	Dolomitic Lime	20,412		80%	0.8700	4,843	17,758	
1998	High-Calcium Lime	95,522		80%	0.7500	19,539	71,642	✓ Default Production Data
	Dolomitic Lime	19,478		80%	0.8700	4,622	16,946	
1999	High-Calcium Lime	90,602		80%	0.7500	18,532	67,952	✓ Default Production Data

Required Data Input Cells

Control / Cement / Lime / Limestone / Soda Ash / Iron & Steel / Ammonia & Urea / Nitric / Adipic / Aluminum / HCFC-22 / QDS / Semi

Step (5) Enter Emission Factors and Activity Data for Limestone and Dolomite Use

Control Worksheet

The next inputs on the control worksheet are emission factors for limestone and dolomite use, and magnesium produced from dolomite. Limestone (CaCO₃) and dolomite (CaMg(CO₃)₂) are basic raw materials used by a wide variety of industries, including the construction, agriculture, chemical, glass manufacturing, environmental pollution control, and metallurgical industries such as magnesium (Mg) production.

Limestone and Dolomite Use Sector Worksheet

Production data for limestone and dolomite use, and magnesium production from dolomite are required as inputs in the blue cells of the limestone and dolomite worksheet as displayed in Figure 6. As an example, CO₂ is emitted as a by-product from the reaction of limestone or dolomite with impurities in the iron ore and fuels heated in a blast furnace. Activity data for limestone and dolomite use by state is available from USGS (2026c).

The quantities of limestone consumed for industrial purposes, dolomite consumed for industrial purposes, and magnesium produced from dolomite are multiplied by their respective emission factors. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E). For default data, each state's total limestone consumption (as reported by USGS) is multiplied by the ratio of national limestone consumption for industrial uses to

total National limestone consumption. Equation 4 shows this calculation for CO₂ emissions from limestone and dolomite use.

Equation 4. Emission Equation for Limestone and Dolomite Use

$$\text{Emissions (MTCO}_2\text{E)} = \text{Consumption (metric tons)} \times \text{Emission Factor (MT CO}_2\text{/MT production)}$$

Figure 6. Example of Activity Data Applied in the Limestone and Dolomite Use Worksheet

State Inventory Tool - Industrial Processes Module

4. Limestone and Dolomite Use in Colorado

Click here to find where these data are available.

Emissions from limestone and dolomite use result from industrial consumption. The quantities of limestone consumed for industrial purposes, dolomite consumed for industrial purposes, and magnesium produced from dolomite are multiplied by their respective emission factors. Industrial uses include the consumption of limestone and dolomite for flux stone production, glass manufacturing, flue gas desulfurization (FGD), Mg production through the thermic reduction of dolomite, chemical stone manufacturing, mine dusting or acid water treatment, acid neutralization, and sugar refining. The emissions are then converted from metric tons of carbon equivalents (MTC) to metric tons of carbon dioxide equivalents (MTCO₂E). For default data, each state's total limestone consumption (as reported by USGS) is multiplied by the ratio of national limestone consumption for industrial uses to total national limestone consumption. Additional information on these calculations, including a definition of industrial uses, is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet
Check All Boxes
Clear All Data

	Consumption (Metric Tons)	Emission Factor (t CO ₂ /t production)	Emissions (MTC)	Emissions (MTCO ₂ E)	
1990 Limestone	-	0.4400	=	-	☒ Default Production Data?
1990 Dolomite	-	0.4840	=	-	☒ Default Production Data?
1990 Magnesium Production from Dolomite	-	1.7967	=	-	
1991 Limestone	-	0.4400	=	-	☒ Default Production Data?
1991 Dolomite	-	0.4840	=	-	☒ Default Production Data?
1991 Magnesium Production from Dolomite	-	1.7967	=	-	
1992 Limestone	-	0.4400	=	-	☒ Default Production Data?
1992 Dolomite	-	0.4840	=	-	☒ Default Production Data?
1992 Magnesium Production from Dolomite	-	1.7967	=	-	

Required Data Input Cells

Step (6) Enter Emission Factors and Activity Data for Soda Ash Manufacture and Consumption

Control Worksheet

Soda ash manufacture and consumption emission factors are required inputs on the control worksheet in order to calculate emissions from this source. Although only three states produced soda ash at the time of publication (Wyoming, California, and Colorado), all states consumed it. Thus, all states should estimate CO₂ emissions from soda ash consumption.

Soda Ash Manufacture and Consumption Sector Worksheet

Production data for the manufacture and consumption of soda ash are required as inputs in the blue cells of the soda ash manufacture worksheet as shown in Figure 7. Under the soda ash production method used in some states, trona (an ore from which natural soda ash is made) is calcined in a rotary kiln and chemically transformed into a crude soda ash that requires further processing. CO₂ and water are generated as a by-product of the calcination process. CO₂ is also released when soda ash is

consumed in products such as glass, soap, and detergents. Activity data for soda ash manufacture and consumption is available from USGS (2024b).

Emissions from soda ash manufacture and consumption are calculated by multiplying the quantity of soda ash manufactured (Wyoming only) and the quantity of soda ash consumed by their respective emission factors. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E) as shown in Equation 5.

Equation 5. Emission Equation for Soda Ash Manufacture and Consumption

$$\text{Emissions (MTCO}_2\text{E)} = \text{Manufacture/Consumption (metric tons)} \times \text{Emission Factor (MT CO}_2\text{/MT production)}$$

Figure 7. Example of Activity Data Applied in the Soda Ash Manufacture and Consumption Worksheet

		Manufacture and Consumption (Metric Tons)	Emission Factor (t CO ₂ /t production)	Emissions (MTCE)	Emissions (MTCO ₂ E)	
1990	Manufacture	-	0.0974	-	-	☒ Default Production Data?
	Consumption	86,482	0.4150	3,788	35,890	☒ Default Consumption Data?
1991	Manufacture	-	0.0974	-	-	☒ Default Production Data?
	Consumption	83,671	0.4150	3,472	33,600	☒ Default Consumption Data?
1992	Manufacture	-	0.0974	-	-	☒ Default Production Data?
	Consumption	85,744	0.4150	3,559	34,580	☒ Default Consumption Data?
1993	Manufacture	-	0.0974	-	-	☒ Default Production Data?
	Consumption	86,482	0.4150	3,788	36,001	☒ Default Consumption Data?

Step (7) Enter Emission Factors and Activity Data for Iron and Steel Production

Control Worksheet

Emission factors for the following iron and steel production processes are required as inputs on the control worksheet: Basic Oxygen Furnace (BOF) at Integrated Mill with Coke Ovens, Basic Oxygen Furnace (BOF) at Integrated Mill without Coke Ovens, Electric Arc Furnace (EAF), and open hearth furnace (OHF). In addition to being an energy intensive process, the production of iron and steel also generates process-related emissions of CO₂. It is strongly advised that users enter state-specific information, as default data are based on national averages and are not available for all years.

Iron and Steel Production Sector Worksheet

Activity data for the production of iron and steel are required as inputs in the blue cells of the iron and steel worksheet displayed in Figure 8. The basic activity data needed are the quantities of crude steel produced (defined as first cast product suitable for sale or further processing) by production method. It is strongly advised that users enter state-specific information, as default data are based on national averages, are not available for all years, and are likely to be inaccurate for states. The national data are provided by the American Iron and Steel (AIS) Annual Statistics Report 2010 (AIS 2011).

Emissions from iron and steel production are based on the state-level production data assigned to production method based on the national distribution of production by method. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E) as shown in Equation 6.

Equation 6. Emission Equation for Iron and Steel Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Manufacture/Consumption (metric tons)} \times \text{Emission Factor (MT CO}_2\text{/MT production)}$$

Figure 8. Example of Activity Data Applied in the Iron and Steel Production Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options Type a question for help

6. Iron and Steel Production in Colorado

Click here to find where these data are available.

Iron and steel production generate process-related emissions. The basic activity data needed are the quantities of crude steel produced (defined as first cast product suitable for sale or further processing) by production method. Default values are based on the state-level production data assigned to production method based on the national distribution of production by method. **It is strongly advised that users enter state-specific information, as default data are based on national averages, are not available for all years, and are likely to be inaccurate for states.** Activity data are then multiplied by the appropriate emission factor. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E). This methodology is based on the Draft 2006 IPCC Guidelines for National GHG Inventories.

Return to Control Sheet

Check All Boxes

Clear All Data

	Production Method	State Production (Metric Tons)	Emission Factor (t CO ₂ /t production)	Emissions (MTCE)	Emissions (MTCO ₂ E)
1997	BOF with coke ovens	245,436	1.72	115,132	422,150
	BOF without coke ovens	245,436	1.46	97,728	358,337
	SAF	381,968	0.08	8,334	30,558
	OHF	-	1.72	-	-
	Total				811,044
1998	BOF with coke ovens	225,576	1.72	105,816	387,991
	BOF without coke ovens	225,576	1.46	89,820	329,342
	SAF	370,888	0.08	8,092	29,671
	OHF	-	1.72	-	-
	Total			203,728	747,004

Default Production Data?

Default Production Data?

Required Data Input Cells

Step (8) Enter Emission Factors and Activity Data for Ammonia Manufacture

Control Worksheet

The emission factors for ammonia and urea production are the next required inputs on the control worksheet. Emissions of CO₂ occur during the production of synthetic ammonia, primarily through the use of natural gas as a feedstock.

Ammonia Manufacture Sector Worksheet

Data for the production of ammonia and consumption of urea are required inputs in the blue cells on the ammonia production worksheet, shown in Figure 9. Activity data for ammonia manufacture by state is available from USGS (2026d). Activity data for urea consumption by state is estimated based on state data from AAPFCO (2017) and TVA (1991 through 1994).

Emissions from ammonia production and urea consumption are calculated by multiplying the quantity of ammonia produced and urea applied by their respective emission factors. Emissions from urea consumption are subtracted from emissions due to ammonia production. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E) as shown in Equation 7 and Equation 8.

Equation 7. Emission Equation for Ammonia Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Production of Ammonia (metric tons)} \times \text{Emission Factor (MT CO}_2\text{/MT activity)} - \text{Emissions from Urea (MTCO}_2\text{E)}$$

Equation 8. Emission Equation for Urea Consumption

$$\text{Emissions (MTCO}_2\text{E)} = \text{Consumption of Urea (metric tons)} \times \text{Emission Factor (MT CO}_2\text{/MT activity)}$$

Figure 9. Example of Activity Data Applied in the Ammonia Production and Urea Consumption Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options

7. Ammonia Production and Urea Consumption in Colorado

Click here to find where these data are available.

Emissions from ammonia production and urea application are calculated by multiplying the quantity of ammonia produced and urea applied by their respective emission factors. Emissions from urea application are subtracted from emissions due to ammonia production. The emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet
Check All Boxes
Clear All Data

	Production & Consumption (Metric Tons)	Emission Factor (mt CO ₂ /mt activity)	Subtract emissions from Urea	Emissions (MTCE)	Emissions (MTCO ₂ E)	
1990 Ammonia Production	-	1.2	3,071	-	-	☒ Default Production Data?
Urea Consumption	4,206	0.73		837	3,071	
1991 Ammonia Production	-	1.2	2,683	-	-	☒ Default Production Data?
Urea Consumption	3,676	0.73		2,683	2,683	
1992 Ammonia Production	-	1.2		-	-	☒ Default Production Data?
Urea Consumption	3,525	0.73		2,573	2,573	
1993 Ammonia Production	-	1.2	3,163	-	-	☒ Default Production Data?
Urea Consumption	4,332	0.73		863	3,163	
1994 Ammonia Production	-	1.2	2,871	-	-	☒ Default Production Data?
Urea Consumption	3,933	0.73		783	2,871	

Required Data Input Cells

Step (9) Enter Emission Factors and Activity Data for Aluminum Production

Control Worksheet

The emission factors for aluminum production are the next required inputs on the control worksheet. The aluminum production industry is thought to be the largest source of two PFCs – tetrafluoromethane (CF₄) and hexafluoroethane (C₂F₆). Emissions of these two potent greenhouse gases (GHGs) and CO₂ occur during the reduction of alumina in the primary smelting process.² Emission factors are required as inputs on the control worksheet for PFC emissions, CO₂ emissions from Prebake technology, and CO₂ emissions from Söderberg technology.

Aluminum Production Sector Worksheet

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). Data for the production of aluminum are required in the blue cells of the aluminum worksheet, shown in Figure 10. Activity data for aluminum production is available from USGS (2024c).

PFC emissions from aluminum production are calculated by multiplying the quantity of aluminum produced during a year by the specific emission factor for that year. CO₂ emissions from aluminum production are calculated by multiplying the quantity of aluminum produced during a year by a weighted CO₂ emission factor. The emission factor is weighted by the percent of aluminum production using either Prebake or Söderberg technology. If the percent of production by technology type is unknown, a default percentage is assumed based on national data from the U.S. Inventory (U.S. EPA 2025). It is strongly advised that users enter state-specific information, as default data are based on national averages.

These emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E), shown in Equation 11.

Equation 9. Emission Equation for Aluminum Production

$$\begin{aligned}
 \text{Total Emissions (MTCO}_2\text{E)} &= \text{PFC Emissions (MTCO}_2\text{E)} + \text{CO}_2 \text{ Emissions (MTCO}_2\text{E)} \\
 \text{PFC Emissions (MTCO}_2\text{E)} &= \\
 &\text{Production of Aluminum (metric tons) x Emission Factor (MT CE/MT production)} \\
 \text{CO}_2 \text{ Emissions (MTCO}_2\text{E)} &= \\
 &\text{Production of Aluminum (metric tons) x } [(\text{Percent of Production}_{\text{Prebake}} \times \text{EF}_{\text{Prebake}}) + (\text{Percent of} \\
 &\quad \text{Production}_{\text{Söderberg}} \times \text{EF}_{\text{Söderberg}})] \text{ (MT CE/MT production)}
 \end{aligned}$$

² Perfluorinated carbons are not emitted during the smelting of recycled aluminum.

Figure 10. Example of Activity Data Applied in the Aluminum Production Worksheet

11. Colorado Aluminum Production

Click here to find where these data are available.

Aluminum production generates process-related emissions. The basic activity data needed are the quantities of aluminum produced and percentages of aluminum produced by technology type. Default percentages of aluminum production by technology type are based on the national distribution of production by technology. It is strongly advised that users enter state-specific information, as default data are based on national averages, are not available for all years, and are likely to be inaccurate for states.

Emissions from aluminum production are then calculated by multiplying the quantity of aluminum produced during a year by the PFC emission factor and the CO₂ emission factors for that year. These emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Check All Boxes

Clear All Data

Default Technology Share of Soderberg and Prebake Facilities

Year	Production (Metric Tons)	PFC Emission Factor (t CER production)	Soderberg Facilities %	Soderberg CO ₂ Emission Factor (t CER production)	Prebake Facilities %	Prebake CO ₂ Emission Factor (t CER production)	PFC Emissions (MTCE)	Carbon Emissions (MTCE)	Total Emissions (MTCO ₂ E)
1990	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1991	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1992	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1993	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1994	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1995	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1996	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1997	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1998	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
1999	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
2000	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
2001	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	
2002	x (0.4255)	x (0.4636)	x 100.00%	x (0.4364)	=	-	-	-	

Default Production Data?

Option to Use Default Technology Share

Required Data Input Cells

Step (10) Enter Emission Factors and Activity Data for Nitric Acid Production

Control Worksheet

The emission factor for nitric acid production is the next required input for the control worksheet. The production of nitric acid (HNO₃) produces N₂O as a by-product, via the oxidation of ammonia. Nitric acid is a raw material used primarily to make synthetic commercial fertilizer and is also a major component in the production of adipic acid (a feedstock for nylon) and explosives.

Nitric Acid Production

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). Data for the amount of nitric acid produced, as well as the percent N₂O released after pollution control are inputs for the nitric acid worksheet as seen in Figure 11. Activity data for nitric acid production is available from SRI 2000. The production of nitric acid (HNO₃) produces N₂O as a by-product, via the oxidation of ammonia. During this reaction, N₂O is formed as a by-product and is released from reactor vents into the atmosphere. At present, the nitric industry controls for oxides of nitrogen through two technologies: non-selective catalytic reduction (NSCR) and selective catalytic reduction (SCR). Only one of these technologies, NSCR, is effective at destroying N₂O emissions in the process of destroying NO_x emissions.

Emissions from nitric acid production are calculated by multiplying the quantity of nitric acid produced by an emission factor and by the percentage of N₂O released after pollution controls are considered. These emissions are then converted from metric tons of carbon equivalents (MTCE) to metric tons of carbon dioxide equivalents (MTCO₂E) as seen in Equation 9.

Equation 10. Emission Equation for Nitric Acid Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Production of Nitric Acid (metric tons)} \times \text{Emission Factor (MT N}_2\text{O/MT production)} \times \text{Percent N}_2\text{O Released after Pollution Control} \times \text{GWP N}_2\text{O}$$

Figure 11. Example of Activity Data Applied in the Nitric Acid Production Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options

8. Nitric Acid Production in Colorado

Click here to find where these data are available.

Emissions from nitric acid production are calculated by multiplying the quantity of nitric acid produced by an emission factor and by the percentage of N₂O released after pollution controls are taken into account. These emissions are then converted from metric tons of N₂O to metric tons of carbon equivalents (MTCE) and metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Clear All Data

Use Default Pollution Control Factor (100%, no pollution control)

	Production (Metric Tons)	Emission Factor (t N ₂ O/t production)	Percent N ₂ O Released after Pollution Control	Emissions (Metric Tons N ₂ O)	Emissions (MTCE)	Emissions (MTCO ₂ E)
1990		0.0080	100%	-	-	-
1991		0.0080	100%	-	-	-
1992		0.0080	100%	-	-	-
1993				-	-	-
1994				-	-	-
1995		0.0080	100%	-	-	-
1996		0.0080	100%	-	-	-

There is no default data associated with this sector. You will need to fill in all required production data in order to estimate emissions.

Required Data Input Cells

Step (11) Enter Emission Factors and Activity Data for Adipic Acid Production**Control Worksheet**

The emission factor for adipic acid production is required next on the control worksheet. About 90 percent of all adipic acid produced in the United States is used in the production of nylon 6,6, as well as production of some low-temperature lubricants. It is also used to provide foods with a “tangy” flavor.

Adipic Acid Production Sector Worksheet

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). Data for the amount of adipic acid produced, as well as the percent N₂O released after pollution control are inputs for the adipic acid worksheet as seen in Figure 12. A dialogue box will appear if adipic acid is not produced in your state. Note that plants may consider this data confidential and could be reluctant to disclose it, in which case, states should use the adipic acid production capacity data in the Chemical Market Reporter.

Adipic acid is produced through a two-stage process. The first stage involves the oxidation of cyclohexane to form a cyclohexanone/cyclohexanol mixture. The second stage involves the oxidation of ketone-alcohol with nitric acid. N₂O is generated as a by-product of this reaction and enters the waste

gas stream. In the United States, this waste gas is treated to remove NO_x and other regulated pollutants (and, in some cases, N₂O as well) and is then released into the atmosphere.

Emissions from adipic acid production are calculated by multiplying the quantity adipic acid produced by an emission factor and by the percentage of N₂O released after pollution controls are considered. These emissions are then converted from metric tons of N₂O to metric tons of carbon equivalents (MTCE) and then metric tons of carbon dioxide equivalents (MTCO₂E), shown in Equation 10.

Equation 11. Emission Equation for Adipic Acid Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Production of Adipic Acid (metric tons)} \times \text{Emission Factor (MT N}_2\text{O/MT production)} \times \text{Percent N}_2\text{O Released after Pollution Control} \times \text{GWP N}_2\text{O}$$

Figure 12. Example of Activity Data Applied in the Adipic Acid Production Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options Type

9. Adipic Acid Production in Colorado

Click here to find where these data are available.

Emissions from adipic acid production are calculated by multiplying the quantity adipic acid produced by an emission factor and by the percentage of N₂O released after pollution controls are taken into account. These emissions are then converted from metric tons of N₂O to metric tons of carbon equivalents (MTCE) and metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Clear All Data

Use Default Pollution Control Factor (100%, no pollution control)

	Production (Metric Tons)	Emission Factor (t N ₂ O/t production)	Percent N ₂ O Released after Pollution Control	Emissions (Metric Tons N ₂ O)	Emissions (MTCE)	Emissions (MTCO ₂ E)
1990		0.3000	100%	-	-	-
1991		0.3000	100%	-	-	-
1992		0.3000	100%	-	-	-
1993			100%	-	-	-
1994			100%	-	-	-
1995		0.3000	100%	-	-	-

There is no default data associated with this sector. You will need to fill in all required production data in order to estimate

Required Data Input Cells

Step (12) Enter Emission Factors and Activity Data for HCFC-22 Production

Control Worksheet

The next emission factor required for the control worksheet is for HFC-23 emissions resulting from HCFC-22 production, in metric tons of HFC-23 emitted per metric ton of HCFC-22 produced.

HCFC-22 Production Sector Worksheet

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). HFC-23, one type of HFC, is known to be emitted in significant quantities as a by-product of HCFC-22 production. Data for the production of HCFC-22 are required inputs in the blue cells on the HCFC-22 production worksheet, shown in Figure 13. In order to obtain activity data, in-state manufacturers of HCFC-22 should be consulted first. Additionally, the Chemical Manufacturers Association (Washington, D.C.), Alliance for Responsible CFC Policy (Arlington, VA), and Grant Thornton Consulting (Washington, D.C.) can be contacted for information on state-by-state production numbers.

Emissions from HCFC-22 production are calculated by multiplying the quantity of HCFC-22 produced by an emission factor. The emissions are then converted from metric tons of HFC-23 to metric tons of carbon equivalents (MTCE) and then metric tons of carbon dioxide equivalents (MTCO₂E) as in Equation 12.

Equation 12. Emission Equation for HCFC-22 Production

$$\text{Emissions (MTCO}_2\text{E)} = \text{Production of HCFC-22 (metric tons)} \times \text{Emission Factor (MT HFC-23/MT production)} \times \text{GWP of HFC-23}$$

Figure 13. Example of Activity Data Applied in the HCFC-22 Production Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options

11. HCFC-22 Production in Colorado

Click here to find where these data are available.

Emissions from HCFC-22 production are calculated by multiplying the quantity of HCFC-22 produced by an emission factor. The emissions are then converted from metric tons of HFC-23 to metric tons of carbon equivalents (MTCE) and metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Clear All Data

	Production (Metric Tons)	Emission Factor (t HFC-23/t production)	Emissions (Metric Tons HFC-23)	Emissions (MTCE)	Emissions (MTCO ₂ E)
1990		0.0200	-	-	-
1991		0.0200	-	-	-
1992		0.0200	-	-	-
1993		0.0200	-	-	-
1994		0.0200	-	-	-
1995		0.0200	-	-	-
1996		0.0200	-	-	-
1997		0.0200	-	-	-
1998		0.0200	-	-	-

There is no default data associated with this sector. You will need to fill in all required production data in order to estimate emissions.

Required Data Input Cells

Step (13) Review Sector Worksheet for Consumption of Substitutes for Ozone-Depleting Substances (ODS)

Control Worksheet

There are no emission factor inputs required for the consumption of substitutes for ozone-depleting substances (ODS) as state-apportioned emissions are provided on the sector-specific worksheet. Hydrofluorocarbons (HFCs) are used primarily as alternatives to several classes of ODS that are being phased out under the terms of the Montreal Protocol and the Clean Air Act Amendments of 1990. ODSs, which include chlorofluorocarbons (CFCs), halons, carbon tetrachloride, methyl chloroform, and hydrochlorofluorocarbons (HCFCs), are used in a variety of industrial applications including refrigeration and air conditioning equipment, aerosols, solvent cleaning, fire extinguishing, foam blowing, and sterilization. Although their substitutes, HFCs, are not harmful to the stratospheric ozone layer, they are powerful GHGs.

Consumption of ODS Sector Worksheet

There are no inputs required for this worksheet, though you are able to input your own emissions estimates in the blue cells. You should review this worksheet to learn your state's contribution to emissions resulting from the consumption of ODS substitutes. The major end uses that consume substitutes for ozone-depleting substances include motor vehicle air conditioning, commercial and industrial refrigeration and air conditioning, residential refrigeration and air conditioning, aerosols, solvent cleaning, fire extinguishing equipment, foam production, and sterilization.

Emissions of HFCs, PFCs, and SF₆ from ODS substitute production are sourced directly from the U.S. Inventory's state-level emission estimates (U.S. EPA 2025). The U.S. Inventory estimates emissions by apportioning regional HFC emissions estimates to each state based on population, as outlined in the *Methodology Report for Inventory of US Greenhouse Gas Emissions and Sinks by State: 1990-2022*, available here: https://www.epa.gov/system/files/documents/2024-09/method-report_ghgibystate-508-september.pdf (U.S. EPA 2024).

For the U.S. Inventory's state disaggregation, annual emissions per capita for the six regions analyzed in Hu et al. (2017) were used. Specifically, emissions for HFC-125, HFC-134a, and HFC-143a from 2008–2014 and emissions for HFC-32 from 2010–2014 were available. Because Hu et al. (2017) estimates cover the 48 contiguous States and the District of Columbia, emissions estimates from the remaining States (Alaska and Hawaii) were derived strictly based on the State's or territory's population compared to the national population for the full time series 1990-2023. Likewise, the emissions of HFCs other than the four listed above were distributed to all states and territories by population. The emissions of HFC-32, HFC-125, HFC-134a, and HFC-143a were distributed to the six regions in the same ratio as the best estimate of such distribution shown in Hu et al. (2017).

Step (14) Review Sector Worksheet for Semiconductor Manufacture

Control Worksheet

There are no emission factor inputs required for semiconductor manufacture on the control worksheet as the calculations for this sector are performed on the sector-specific worksheet. The semiconductor industry employs multiple long-lived fluorinated gases in the plasma etching and chemical vapor deposition processes. These include the PFCs CF₄, C₂F₆, and C₃F₈; HFC-23; NF₃, and SF₆. With present industry growth and the increasing complexity of microchips, emissions from the semiconductor industry are expected to increase significantly.

Semiconductor Manufacture Sector Worksheet

There are no inputs required for this worksheet. You should review this worksheet to learn your state's contribution to emissions resulting from the manufacture of semiconductors. The semiconductor industry employs multiple long-lived fluorinated gases in the plasma etching and chemical vapor deposition processes and include PFCs CF₄, C₂F₆, and C₃F₈; HFC-23; NF₃, and SF₆.

Emissions of HFCs, PFCs, NF₃, and SF₆ from semiconductor production are estimated by apportioning national emissions to each state. National emissions are multiplied by a ratio of the value of a state's semiconductor shipments, as found in U.S. Census Bureau (1997, 2002, 2007, 2012, 2017, and 2023), to the value of national semiconductor shipments. The resulting state emissions are then converted into metric tons of CO₂ equivalents (MTCO₂E) as shown in Equation 14.

Equation 13. Emission Equation for Apportioning Emissions from Semiconductor Manufacture

$$\text{Emissions (MTCO}_2\text{E)} = \frac{\text{National Semiconductor Manufacture Emissions (MTCO}_2\text{E)} \times [\text{Value of State Semiconductor Shipments} / \text{Value of National Semiconductor Shipments}]$$

Step (15) Enter Emission Factors and Activity Data for Electric Power Transmission and Distribution

Control Worksheet

The emission factor for electric power transmission and distribution is required on the control worksheet. The largest use for SF₆, both domestically and internationally, is as an electrical insulator in electricity transmission and distribution equipment, such as gas-insulated high-voltage circuit breakers, substations, transformers, and transmission lines.

Electric Power Transmission and Distribution Sector Worksheet

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). This worksheet requires inputs for the

amount of SF₆ consumed for electric power transmission and distribution as shown in Figure 14. Activity data for electric transmission and distribution are available from U.S. EPA (2025) and EIA (2025).

The largest use for SF₆, both domestically and internationally, is as an electrical insulator in electricity transmission and distribution equipment, such as gas-insulated high-voltage circuit breakers, substations, transformers, and transmission lines. The electric utility industry uses the gas because of its high dielectric strength and arc-quenching abilities. Not all of the electric utilities in the United States use SF₆; use of the gas is more common in urban areas where the space occupied by electrical distribution and transmission facilities is more valuable.

Emissions from electric power transmission and distribution are calculated by multiplying the quantity of SF₆ consumed by an emission factor. The resulting emissions are then converted from metric tons of SF₆ to metric tons of carbon dioxide equivalents (MTCO₂E) as shown in Equation 15. The default assumption is that the emission factor is 1, i.e. all SF₆ consumed is used to replace SF₆ that was emitted. Default activity data for this sector equals national SF₆ emissions apportioned by state electricity sales divided by national electricity sales.

Equation 14. Emission Equation for Electric Power Transmission and Distribution

$$\text{Emissions (MTCO}_2\text{E)} = \text{SF}_6 \text{ Consumption (metric tons SF}_6\text{)} \times \text{Emission Factor (MT SF}_6\text{/MT Consumption)} \times \text{GWP of SF}_6$$

Figure 14. Example of Activity Data Applied in the Electric Power Transmission and Distribution Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options Type a question

14. Electric Power Transmission and Distribution in Colorado

Click here to find where these data are available.

Emissions from electric power transmission and distribution are calculated by multiplying the quantity of SF₆ consumed by an emission factor. The resulting emissions are then converted from metric tons of SF₆ to metric tons of carbon equivalents (MTCE) and metric tons of carbon dioxide equivalents (MTCO₂E). The default assumption is that the emission factor is 1, i.e. all SF₆ consumed is used to replace SF₆ that was emitted. Default activity data for this sector equals national SF₆ emissions apportioned by state electricity sales divided by national electricity sales. Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet

Check All Boxes

Clear All Data

	SF ₆ Consumption (Metric Tons)	Emission Factor (t SF ₆ /t Consumption)	Emissions (Metric Tons SF ₆)	Emissions (MTCE)	Emissions (MTCO ₂ E)	
1990	13	1.0	13	84,021	308,076	☒ Default SF ₆ Consumption Data?
1991	12	1.0	12	80,779	296,191	☒ Default SF ₆ Consumption Data?
1992	12	1.0	12	81,383	298,405	☒ Default SF ₆ Consumption Data?
1993	12	1.0	12	78,908	289,329	☒ Default SF ₆ Consumption Data?
1994	12	1.0	12	75,862	278,159	☒ Default SF ₆ Consumption Data?
1995	11	1.0	11	69,689	255,526	☒ Default SF ₆ Consumption Data?

Required Data Input Cells

Step (16) Enter Emission Factors and Activity Data for Magnesium Production and Processing

Control Worksheet

Primary and secondary production, as well as casting emission factors for magnesium production and processing are required in the control worksheet. The Mg metal production and casting industry uses SF₆ as a cover gas to prevent the violent oxidation of molten Mg in the presence of air.

Magnesium Production and Processing Sector Worksheet

(Note: Emissions in this tool are estimated using IPCC AR5 GWPs. If users are estimating emissions with non-AR5 GWPs, users must enter their own emissions estimates). In the blue input cells on the magnesium worksheet, enter the quantity of primary magnesium produced, secondary magnesium produced, and magnesium cast during a given year as shown in Figure 15. Activity data for the production and processing of magnesium are available at USGS (2024a).

The Mg metal production and casting industry uses SF₆ as a cover gas to prevent the violent oxidation of molten Mg in the presence of air. A gas mixture consisting of CO₂, air, and a small concentration of SF₆ is blown over the molten Mg metal to induce the formation of a protective crust. Most producers of primary Mg metal and most Mg part casters use this technique. SF₆ replaced the previously used sulfur dioxide due to the numerous health and safety risks associated with sulfur dioxide.

Emissions from magnesium production and processing are emitted during the production of primary magnesium, production of secondary magnesium, and casting of magnesium. The emissions are calculated by multiplying the quantity of primary magnesium produced, secondary magnesium produced, and magnesium cast during a given year by their respective emission factors for the same year. The resulting emissions are then converted from metric tons of SF₆ to metric tons of carbon dioxide equivalents (MTCO₂E) as shown in Equation 16.

Equation 15. Emission Equation for Magnesium Production and Processing

$$\text{Emissions (MTCO}_2\text{E)} = \text{Quantity of Magnesium Produced (metric tons)} \times \text{Emission Factor (MT SF}_6\text{/MT Magnesium)} \times \text{GWP of SF}_6$$

Figure 15. Example of Activity Data Applied in the Magnesium Production and Processing Worksheet

State Inventory Tool - Industrial Processes Module

File Edit Module Options Type

15. Magnesium Production and Processing in Colorado

Click here to find where these data are available.

Emissions from magnesium production and processing are emitted during the production of primary magnesium, production of secondary magnesium, and casting of magnesium. The emissions are calculated by multiplying the quantity of primary magnesium produced, secondary magnesium produced, and magnesium cast during a given year by their respective emission factors for the same year. The resulting emissions are then converted from metric tons of SF₆ to metric tons of carbon equivalents (MTCE) and metric tons of carbon dioxide equivalents (MTCO₂E). Additional information on these calculations is available in the Industrial Processes Chapter of the User's Guide.

Return to Control Sheet
Clear All Data

	Magnesium Production and Processing (Metric Tons)	Emission Factor (t SF ₆ /t Magnesium)	Emissions (Metric Tons SF ₆)	Emissions (MTCE)	Emissions (MTCO ₂ E)
1990	Primary Production	0.0012	-	-	-
	Secondary Production	0.0010	-	-	-
	Casting	0.0041	-	-	-
1991	Primary Production	0.0012	-	-	-
	Secondary Production	0.0010	-	-	-
	Casting	0.0041	-	-	-
1992	Primary Production	0.0012	-	-	-
	Secondary Production	0.0010	-	-	-
	Casting	0.0041	-	-	-

Required Data Input Cells

Step (17) Review Summary Information

The steps above provide estimates of total CO₂, N₂O, and HFC, NF₃, PFC, and SF₆ emissions from each IP sector. Total emissions are equal to sum of emissions from each of the fifteen IP sectors, for each year. The information is collected by sector on the summary worksheets. There is a summary worksheet in the IP module that displays results in MMTCO₂E. Additionally, the summary worksheet provides an overview of sources excluded from the current emission estimates. Users should review this list to see if they wish to go back and enter data for any of the omitted IP sectors. Figure 16 shows the summary worksheet that sums the emissions from all sectors in the IP module.

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